

Acknowledgement—The authors wish to thank Professor N. Morita of Toyama University for a sample of pectolinarigenin.

REFERENCES

1. Kimura, K., Okuda, T. and Takano, T. (1963) *Yakugaku Zasshi* **83**, 635, and references cited therein.
2. Inouye, H., Okuda, T. and Hirata, Y. (1965) *Tetrahedron Letters* 1261; (1967) *Chem. Pharm. Bull. (Tokyo)* **15**, 786.
3. Inouye, H., Okuda, T. and Hayashi, T. (1971) *Tetrahedron Letters* 3615.
4. Nakaoki, T. and Morita, N. (1955) *Yakugaku Zasshi* **75**, 171.
5. Hiramoto, M. and Watanabe, K. (1939) *Yakugaku Zasshi* **59**, 261.
6. Sastri, V. D. N. and Seshadri, T. R. (1946) *Proc. Ind. Acad. Sci. A* **23**, 273.
7. Seshadri, T. R. and Sharma, P. (1973) *Indian J. Chem.* **11**, 338.
8. Gritsenko, E. N. and Litvinenko, V. I. (1969) *Khim. Prirod. Soed.* **5**, 55.
9. Mears, J. A. and Mabry, T. J. (1972) *Phytochemistry* **11**, 411.
10. Harborne, J. B. (1967) *Comparative Biochemistry of the Flavonoids*, Academic Press, London.
11. Kamerling, J. P., De Bie, M. J. A. and Vliegthart, J. F. G. (1972) *Tetrahedron* **28**, 3037.
12. Birkhofer, L. and Kaiser, Ch. (1962) *Z. Naturforsch.* **17b**, 359.
13. Bridel, M. and Lavielle, R. (1931) *Bull. Soc. Chim. Biol.* **13**, 636.
14. Harborne, J. B. (1967) *Phytochemistry* **6**, 1643.
15. Mabry, T. J., Markham, K. R. and Thomas, M. B. (1970) *The Systematic Identification of Flavonoids*, Springer, Heidelberg.
16. Rodoriguez, E., Carman, N. J. and Mabry, T. J. (1972) *Phytochemistry* **11**, 409.
17. Prepared in the same way as described by Coxon, B. and Fletcher, H. G. Jr. (1961) *J. Org. Chem.* **26**, 2892.

Phytochemistry, 1975, Vol. 14, pp. 1656–1657. Pergamon Press. Printed in England.

PULVERIN, A NEW CHROMONE FROM THE FRUITS OF *NEOCHAMAELEA PULVERULENTA**

ANTONIO G. GONZÁLEZ, BRAULIO M. FRAGA and OLIVA PINO

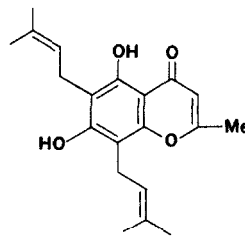
Department of Organic and Biochemistry, University of La Laguna, Instituto de Investigaciones Químicas, C.S.I.C., Tenerife, Spain

(Received 27 January 1975)

Key Word Index—*Neochamaelea pulverulenta*; Cneoraceae; leña santa; chromones; 3,3-dimethylallylspathe-liachromene; pulverin; sitosterol; spatheliabischromene.

Plant. *Neochamaelea pulverulenta* Erndt (Vent) [1] (*Cneorum pulverulentum* Vent.) A voucher specimen is deposited in the Herbarium of the Botanical Department, University of La Laguna. *Source.* Guaza Mountain, Tenerife, Canary Islands, in August. *Uses Medicinal.* *Previous work.* Aerial parts [2,3].

Present work. Green fruits (300 g) were extracted with hot EtOH, concentrated *in vacuo* and chromatographed over SiO₂. Elution with C₆H₆, C₆H₆-EtOAc and EtOAc gave: 3,3-dimethylallylspatheliachromene [4,5], the new natural product pulverin (45 mg), sitosterol and spatheliabischromene [4,5].



(1)

Pulverin. (2-methyl-6,8-di-C-prenyl-5,7-dihydroxychromone) (1), (Found: C, 73.29; H, 7.42; C₂₀H₂₀O₄ requires: C, 73.15; H, 7.37%) mp 147–149°, MS: *m/e* 328 (M⁺), 313, 285, 273, 257, 229, 217 (100%) 205, 177, 128. UV $\lambda_{\max}^{\text{EtOH}}$ 214, 230, 265, 305(sh) nm. IR $\nu_{\max}^{\text{CHCl}_3}$ 3350, 2960, 2900, 2850, 1660, 1600, 1420, 1100, 860 cm⁻¹. NMR (CDCl₃, τ) –2.96 (1H, phenolic proton at C₅), 3.62 (1H,

* Part 6 in the series *Chromenes and Chromones*. For Part V see González, A. G., Fraga, B. M. and Pino, O. (1975) *Rev. Real Acad. Ciencias* **69**, 347.

s, pyronic hydrogen), 4.75 (2H, *m*, 2 C=CH-), 6.55 (4H, *d*, *J* 8 Hz, 2 -CH₂-Ar), 7.66 (3H, *s*, pyronic methyl), 8.18 (6H, *s*, 2 Me-C≡) and 8.25 (6H, *s*, 2 Me-C≡). This compound was synthesized by Seshadri *et al.* [6], mp 149–150°, identical NMR spectrum. The other compounds isolated were identified by comparison with authentic samples.

Acknowledgements—We thank Prof. Primo and J. Sendra (Instituto de Agroquímica, Valencia) for the MS and Prof W. Wildpret and E. Barquín (Botanical Department, University of La Laguna) for supplying us with the material.

REFERENCES

1. Erdmant, G. (1952) *Pollen Morphology and Plant Taxonomy*, Angiosperms, 115, Almqvist-Wiksell, Stockholm.
2. González, A. G., Fraga, B. M. and Castañeda, J. P. (1972) *Anal. Quim.* **68**, 447.
3. González, A. G., Fraga, B. M. and Castañeda, J. P. (1974) *Anal. Quim.* **70**, 452.
4. Taylor, D. R. and Wright, J. A. (1971) *Rev. Latinoam. Quim.* **2**, 84.
5. González, A. G., Fraga, B. M. and Pino, O. (1974) *Phytochemistry* **13**, 2305.
6. Jain, A. C., Lal, P. and Seshadri, T. R. (1969) *Ind. J. Chem.* **7**, 1072.

Phytochemistry, 1975, Vol. 14, pp. 1657–1658. Pergamon Press. Printed in England.

N-METHYL-L-TYROSINE FROM SEEDS OF *COMBRETUM ZEYHERI**

KASUKA MWAULUKA†, E. ARTHUR BELL and BARRY V. CHARLWOOD

Department of Plant Sciences, King's College, 68 Half Moon Lane, London SE24, England
and

JOHN M. BRIGGS

Department of Chemistry, King's College, Strand, London WC2, England

(Received 17 January 1975)

Key Word Index—*Combretum zeyheri*; Combretaceae; *N*-methyl-L-tyrosine; surinamine; new amino acid; andirine; ratanhine; geoffroyine; angeline.

Early reports of the isolation of a nitrogenous compound from the bark of the legumes *Geoffroya surinamensis*, *Ferreira spectabilis* and *Andira anthelmintica* and an American ratanhia extract were summarized by Johnson and Nicolet [2]. The compound was named by various workers, surinamine, geoffroyine, ratanhine, angeline and andirine before the identity of the different isolates was recognized. Goldschmidt [3] confirmed earlier work which indicated that the empirical formula of the compound was C₁₀H₁₃NO₃ and established that his "ratanhine" gave a negative optical rotation in acid solution. Subsequent syntheses of racemic *N*-methyltyrosine [4] and of the *D* and *L* isomers [5, 6] confirmed that the natural compound was *N*-methyltyrosine and that the specific rotation reported by Goldschmidt for

the natural product corresponded to that of the synthetic *D* isomer.

While investigating the free amino acids present in the seeds of an East African plant, *Combretum zeyheri* (Combretaceae), we observed several ninhydrin-reacting "spots" occupying unfamiliar positions on 2D paper chromatograms of the seed extract. One of the more prominent of these gave a brown-purple reaction with ninhydrin and moved slightly faster than tyrosine in both solvents. The compound was isolated by ion exchange chromatography and characterized as *N*-methyl-L-tyrosine, identical in all respects to a synthetic sample.

EXPERIMENTAL

Paper chromatography. Finely ground seed (100 mg) was shaken with 70% EtOH (1 ml) for 2 hr at room temp. After standing for a further 17 hr the suspension was centrifuged and the supernatant liquid used for analyses. 2D chromatograms were descending on Whatman 3 MM paper using 0.1 ml of extract. Solvents used were (1) *n*-BuOH-HOAc-H₂O

* Preliminary account presented at the Phytochemical Society Meeting, Swansea, September, 1974 [1].

† Present address: University of Zambia, P.O. Box 2379, Lusaka, Zambia.